



JEE MAIN 2026
6th APRIL(SHIFT 1)

Mathematics Section A

Q1)

Let $[\cdot]$ denote the greatest integer function. If the domain of the function $f(x) = \sin^{-1}\left(\frac{x + [x]}{3}\right)$ is $[\alpha, \beta)$, then $\alpha^2 + \beta^2$ is equal to:

Options :

- a.
- b.
- c.
- d.

Q2)

Let one root of the quadratic equation in x :

$$(k^2 - 15k + 27)x^2 + 9(k - 1)x + 18 = 0$$

be twice the other. Then the length of the latus rectum of the parabola $y^2 = 6kx$ is equal to:

Options :

- a.

- b. 6
- c. 8
- d. 12

Q3)

Let e_1 and e_2 be two distinct roots of the equation $x^2 - ax + 2 = 0$. Let the sets $\{a \in \mathbb{R} : e_1 \text{ and } e_2 \text{ are the eccentricities of hyperbolas}\} = (\alpha, \beta)$, and $\{a \in \mathbb{R} : e_1 \text{ and } e_2 \text{ are the eccentricities of an ellipse and a hyperbola, respectively}\} = (\gamma, \infty)$. Then $\alpha^2 + \beta^2 + \gamma^2$ is equal to:

Options :

- a. 18
- b. 22
- c. 26
- d. 34

Q4)

Let the set of all values of $k \in \mathbb{R}$ such that the equation $z(\bar{z} + 2 + i) + k(2 + 3i) = 0$, $z \in \mathbb{C}$, has at least one solution, be the interval $[\alpha, \beta]$. Then $9(\alpha + \beta)$ is equal to:

Options :

- a. -10
- b. -8
- c. $10\sqrt{13}$
- d. $8\sqrt{13}$

Q5)

The value of $1^3 - 2^3 + 3^3 - \dots + 15^3$ is:

Options :

- a. 1706
- b. 1856
- c. 1982
- d. 2403

Q6)

The sum of the first ten terms of an A.P. is 160 and the sum of the first two terms of a G.P. is 8. If the first term of the A.P. is equal to the common ratio of the G.P. and the first term of the G.P. is equal to common difference of the A.P., then the sum of all possible values of the first term of the G.P. is:

Options :

a.

$$\frac{34}{9}$$

b.

$$\frac{34}{13}$$

c.

$$\frac{32}{9}$$

d.

$$\frac{32}{13}$$

RAPTORA

Q7)

The number of 4-letter words, with or without meaning, each consisting of two vowels and two consonants that can be formed from the letters of the word INCONSEQUENTIAL, without repeating any letter, is:

Options :

- a. 2670
- b. 2840
- c. 2920
- d. 3600

Q8)

If the coefficients of the middle terms in the binomial expansions of $(1 + \alpha x)^{26}$ and $(1 - \alpha x)^{28}$, $\alpha \neq 0$, are equal, then the value of α is:

Options :

- a. $\frac{1}{13}$
- b. $\frac{14}{13}$
- c. $\frac{27}{7}$
- d. $\frac{7}{27}$

Q9)

A data consists of 20 observations $x_1, x_2, \dots, x_{20}$. If $\sum_{i=1}^{20} (x_i + 5)^2 = 2500$ and $\sum_{i=1}^{20} (x_i - 5)^2 = 100$, then the ratio of mean to standard deviation of this data is:

Options :

- a. 2:1
- b. 3:1
- c. 3:2
- d. 4:1

Q10)

A bag contains $(N + 1)$ coins – N fair coins, and one coin with 'Head' on both sides. A coin is selected at random and tossed. If the probability of getting 'Head' is $\frac{9}{16}$, then N is equal to:

Options :

- a. 5
- b. 7
- c. 8
- d. 9

Q11)

If the eccentricity e of the hyperbola $\frac{x^2}{a^2} - \frac{y^2}{b^2} = 1$, passing through $(6, 4\sqrt{3})$, satisfies $15(e^2 + 1) = 34e$, then the length of the latus rectum of the hyperbola $\frac{x^2}{b^2} - \frac{y^2}{2(a^2 + 1)} = 1$ is:

Options :

- a. 10
- b. 20
- c. 25
- d. 30

Q12)

Let chord PQ of length $3\sqrt{13}$ of the parabola $y^2 = 12x$ be such that the ordinates of points P and Q are in the ratio 1:2. If the chord PQ subtends an angle α at the focus of the parabola, then $\sin \alpha$ is equal to:

Options :

- a. $\frac{3}{5}$
- b. $\frac{4}{5}$

c. $\frac{5}{13}$

d. $\frac{12}{13}$

Q13)

Let $0 < \alpha < 1$, $\beta = \frac{1}{3\alpha}$ and $\tan^{-1}(1-\alpha) + \tan^{-1}(1-\beta) = \frac{\pi}{4}$. Then $6(\alpha + \beta)$ is equal to:

Options :

a. 6

b. 7

c. 8

d. 9

Q14)

Let $S = \{\theta \in (-2\pi, 2\pi) : \cos \theta + 1 = \sqrt{3} \sin \theta\}$.

Then $\sum_{\theta \in S} \theta$ is equal to:

Options :

a. $-\frac{2\pi}{3}$

b. $-\frac{4\pi}{3}$

c. $\frac{2\pi}{3}$

d. $\frac{4\pi}{3}$

Q15)

Let the image of the point $P(1, 6, a)$ in the line $L: \frac{x}{1} = \frac{y-1}{2} = \frac{z-a+1}{b}, b > 0$, be

$\left(\frac{a}{3}, 0, a+c\right)$. If $S(\alpha, \beta, \gamma), \alpha > 0$, is the point on L such that the distance of S from the foot of perpendicular from the point P on L is $2\sqrt{14}$, then $\alpha + \beta + \gamma$ is equal to:

Options :

- a. 19
- b. 20
- c. 21
- d. 22

Q16)

Let a line L be perpendicular to both the lines

$$L_1: \frac{x+1}{3} = \frac{y+3}{5} = \frac{z+5}{7} \text{ and } L_2: \frac{x-2}{1} = \frac{y-4}{4} = \frac{z-6}{7}.$$

If θ is the acute angle between the lines L and

$$L_3: \frac{x-\frac{8}{2}}{2} = \frac{y-\frac{4}{7}}{1} = \frac{z}{2}, \text{ then } \tan \theta \text{ is equal to:}$$

Options :

- a. $\frac{3}{2}\sqrt{2}$
- b. $\frac{5}{2}\sqrt{2}$
- c. $\frac{5}{3}\sqrt{2}$
- d. $\frac{4}{3}\sqrt{2}$

Q17)

The value of $\lim_{x \rightarrow 0} \left(\frac{x^2 \sin^2 x}{x^2 - \sin^2 x} \right)$ is:

Options :

- a. 2
- b. 3
- c. 4
- d. 6

Q18)

The value of the integral $\int_{-\frac{\pi}{4}}^{\frac{\pi}{4}} \left(\frac{32 \cos^4 x}{1 + e^{\sin x}} \right) dx$ is:

Options :

- a. $4\pi + 2$
- b. $3\pi + 8$
- c. $3\pi + 4$
- d. $4\pi + 3$

Q19)

The area of the region $\{(x, y) : 0 \leq y \leq 6 - x, y^2 \geq 4x - 3, x \geq 0\}$ is:

Options :

- a. 8
- b. 9
- c. 12
- d. 15

Q20)

Let e be the base of natural logarithm and let $f : \{1, 2, 3, 4\} \rightarrow \{1, e, e^2, e^3\}$ and $g : \{1, e, e^2, e^3\} \rightarrow \left\{1, \frac{1}{2}, \frac{1}{3}, \frac{1}{4}\right\}$ be two bijective functions such that f is strictly decreasing and g is strictly increasing. If $\phi(x) = \left[f^{-1} \left\{ g^{-1} \left(\frac{1}{2} \right) \right\} \right]^x$, then the area of the region $R = \{(x, y) : x^2 \leq y \leq \phi(x), 0 \leq x \leq 1\}$ is:

Options :

a.

$$\frac{3 - \log_e(2)}{3 \log_e(2)}$$

b.

$$\frac{1}{3 \log_e(2)}$$

c.

$$3 + \log_e(2)$$

d.

$$\frac{3 + \log_e(2)}{2 + \log_e(3)}$$

Mathematics Section B

Q21)

Let $A = \begin{bmatrix} -1 & 1 & -1 \\ 1 & 0 & 1 \\ 0 & 0 & 1 \end{bmatrix}$ satisfy

$$A^2 + \alpha \left(\text{adj} \left(\text{adj} (A) \right) \right) + \beta \left(\text{adj} (A) \left(\text{adj} \left(\text{adj} (A) \right) \right) \right) = \begin{bmatrix} 2 & -2 & 2 \\ -2 & 0 & -1 \\ 0 & 0 & -1 \end{bmatrix} \text{ for}$$

some $\alpha, \beta \in \mathbb{R}$.

Then $(\alpha - \beta)^2$ is equal to _____

Q22)

Let the centre of the circle $x^2 + y^2 + 2gx + 2fy + 25 = 0$ be in the first quadrant and lie on the line $2x - y = 4$. Let the area of an equilateral triangle inscribed in the circle be $27\sqrt{3}$. Then the square of the length of the chord of the circle on the line $x = 1$ is _____.

Q23)

If $\vec{a} = \hat{i} + \hat{j} + \hat{k}$, $\vec{b} = \hat{j} - \hat{k}$ and $\vec{c}$ be three vectors such that $\vec{a} \times \vec{c} = \vec{b}$ and $\vec{a} \cdot \vec{c} = 3$, then $\vec{c} \cdot (\vec{a} - 2\vec{b})$ is equal to _____.

Q24)

For the functions $f(\theta) = \alpha \tan^2 \theta + \beta \cot^2 \theta$, and

$g(\theta) = \alpha \sin^2 \theta + \beta \cos^2 \theta$, $\alpha > \beta > 0$, let $\min_{0 < \theta < \frac{\pi}{2}} f(\theta) = \max_{0 < \theta < \pi} g(\theta)$. If the first

term of a G.P. is $\left(\frac{\alpha}{2\beta}\right)$, its common ratio is $\left(\frac{2\beta}{\alpha}\right)$ and the sum of its first 10

terms is $\frac{m}{n}$, $\gcd(m, n) = 1$, then $m + n$ is equal to _____.

Q25)

Let $y = y(x)$ be the solution of the differential equation

$(x^2 - x\sqrt{x^2 - 1})dy + (y(x - \sqrt{x^2 - 1}) - x)dx = 0$, $x \geq 1$. If $y(1) = 1$, then the

greatest integer less than $y(\sqrt{5})$ is _____.

Physics Section A

Q26)

The density ρ of a uniform cylinder is determined by measuring its mass m , length l and diameter d . The measured values of m , l and d are 97.42 ± 0.02 g, 8.35 ± 0.05 mm and 20.20 ± 0.02 mm, respectively. Calculated percentage fractional error in ρ is _____.

Options :

- a. 0.63%
- b. 0.82%
- c. 0.72%
- d. 0.25%

Q27)

The potential energy of a particle changes with distance x from a fixed origin as $V = \frac{A\sqrt{x}}{x+B}$, where A and B are constant with appropriate dimensions. The dimensions of AB are _____.

Options :

- a. $[M^1 L^{5/2} T^{-2}]$
- b. $[M^{3/2} L^{5/2} T^{-2}]$
- c. $[M^1 L^2 T^{-2}]$
- d. $[M^1 L^{7/2} T^{-2}]$

Q28)

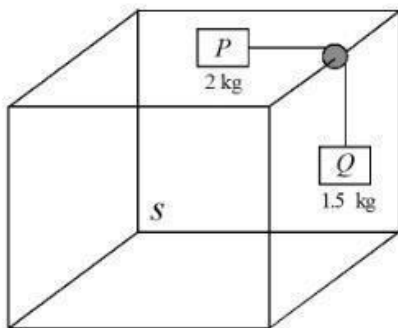
The rain drop of mass 1 g, starts with zero velocity from a height of 1 km. It hits the ground with a speed of 5 m/s. The work done by the unknown resistive force is _____ J.
(take $g = 10 \text{ m/s}^2$)

Options :

- a. - 8.75
- b. - 8.35
- c. - 9.55
- d. - 9.98

Q29)

Two blocks (P and Q) with respectively masses 2 kg and 1.5 kg are joined by a massless thread. These blocks are mounted on a frictionless pulley which is fixed on the edge of a cube (S), as shown in the figure below. Block P is positioned on the top surface which has no friction and block Q is in contact with side-surface, having coefficient friction μ . The cube (S) moves towards the right with acceleration of $\frac{g}{2}$, where g is gravitational acceleration. During this movement the block P and Q remain stationary. The value of μ is _____. (take $g = 10\text{ m/s}^2$)



Options :

- a. 0.33
- b. 0.67
- c. 1
- d. 0.5

Q30)

A lift of mass 1600 kg is supported by thick iron wire. If the maximum stress which the wire can withstand is $4 \times 10^8\text{ N/m}^2$ and its radius is 4 mm , then maximum acceleration the lift can take is _____ m/s^2 . (take $g = 10\text{ m/s}^2$ and $\pi = 3.14$)

Options :

- a. 2.56
- b. 3.89

c. 4.32

d. 5.16

Q31)

A solid sphere of radius 4 cm and mass 5 kg is rotating (rotation axis is passing through the centre of the sphere) with an angular velocity of 1200 rpm. It is brought to rest in 10 s by applying a constant torque. The torque applied and the number of rotations it made before it comes to rest are _____ and _____ respectively.

Options :

a.

$0.128 \pi \text{ Nm}, 100$

b.

$0.0128 \pi \text{ Nm}, 50$

c.

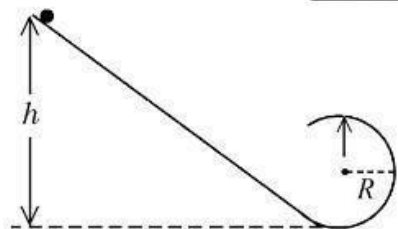
$0.128 \pi \text{ Nm}, 50$

d.

$0.0128 \pi \text{ Nm}, 100$

Q32)

A smooth inclined plane ends in a vertical circular loop, as shown in the figure. A small body is released from height h as shown. If the body exerts a force of three times its weight on the plane at the highest point of circle then the height $h = \alpha R$. The value of α is _____.



Options :

a.

2

b.

4

c.

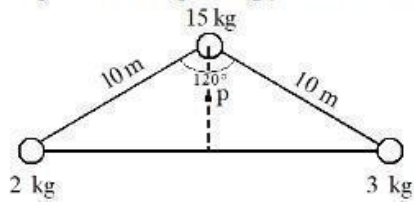
3

d.

6

Q33)

The position of center of mass of three masses 2 kg, 3 kg and 15 kg placed with respect to mid point (p) of normal bisector, as shown in the figure is _____.



Options :

$\left(\frac{\sqrt{3}}{4}, 1.25\right)$

a.

$\left(\frac{\sqrt{3}}{4}, 1.0\right)$

b.

$(0, 0)$

c.

$(1.25, 0)$

d.

Q34)

The two wires A and B of equal cross-section but of different materials are joined together. The ratio of Young's modulus of wire A and wire B is $20/11$. When the joined wire is kept under certain tension the elongations in the wires A and B are equal. If the length of wire A is 2.2 m, then the length of wire B is _____ m.

Options :

1.1

a.

2.22

b.

1.21

c.

4.44

d.

Q35)

Two closed vessels of same volume are joined through a narrow tube and both vessels are filled with air of pressure 90 kPa and temperature 400 K. Keeping the temperature of one vessel constant at 400 K the second vessel temperature is raised to 500 K. The final pressure in the vessels is _____ kPa.

Options :

- a. 100
- b. 120
- c. 90
- d. 105

Q36)

In interference experiment the path difference between two interfering waves at a point A on the screen is $\lambda/3$, where λ is the wavelength of these waves, and at another point B the path difference is $\lambda/6$. The ratio of intensities at points A and B is _____.

Options :

- a. 3
- b. 4
- c. $1/3$
- d. $1/4$

Q37)

A particle is executing simple harmonic motion. Its amplitude is A and time period is 5 sec. The time required by it to move from $x = A$ to $x = \frac{A}{\sqrt{2}}$ is _____ sec.

Options :

- a. $1/4$
- b. $5/4$
- c. $5/8$
- d. $3/8$

Q38)

A thin half ring of radius 35 cm is uniformly charged with a total charge of Q coulomb. If the magnitude of the electric field at centre of the half ring is 100 V/m, then the value of Q is _____ nC.

($\epsilon_0 = 8.85 \times 10^{-12} \text{ C}^2/\text{Nm}^2$ and $\pi = 3.14$)

Options :

a. 2.14

b. 2.44

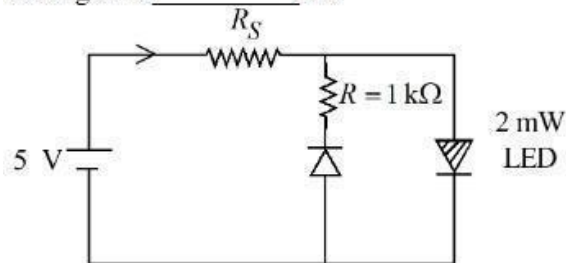
c. 3.25

d. 0.7

Q39)

The maximum rated power of the LED is 2 mW and it is used in the circuit with input voltage of 5 V as shown in the figure below. The current through resistance R_S is 0.5 mA.

The minimum value of the resistance of R_S , to ensure that the LED is not damaged is _____ $\text{k}\Omega$.



Options :

a. 6

b. 2

c. 4

d. 5

Q40)

A point light source emits E.M. waves in free space. A detector, placed at a distance of L m, measures the intensity as I_0 . The detector is now shifted to another location on the same spherical surface ensuring the angle between original location and new location as 45° . The measured intensity at new location will be _____.

Options :

a. $\frac{I_0}{4}$

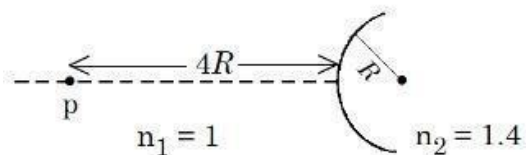
b. I_0

c. $\frac{I_0}{\sqrt{2}}$

d. $\frac{I_0}{2}$

Q41)

A spherical interface lens of radius R separates two media of refractive indices 1 and 1.4 respectively as shown in the figure below. A point source is placed at a distance of $4R$ in front of spherical interface. The magnitude of the magnification of point source image is _____.



Options :

a. 1.66

b. 2.33

c. 2.66

d. 1.33

Q42)

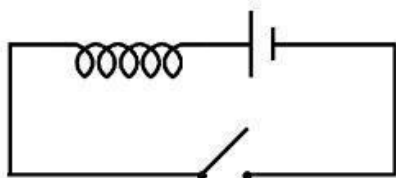
A small cube of side 1 mm is placed at the centre of a circular loop of radius 10 cm carrying a current of 2 A. The magnetic energy stored inside the cube is $\alpha \times 10^{-14}$ J. The value of α is _____.
 ($\mu_0 = 4\pi \times 10^{-7}$ Tm/A, $\pi = 3.14$)

Options :

- a. 6.28
- b. 6.28×10^{-6}
- c. 628
- d. 6.28×10^{-4}

Q43)

An inductor of inductance 10 mH having resistance of 100Ω is connected to battery of E.M.F. 1.0 V through a switch as shown in the figure below. After switch is closed, the ratio of instantaneous voltages across the inductor when the current passing through it is 2 mA and 4 mA is _____.



Options :

- a. 4/3
- b. 3/4
- c. 5/3
- d. 3/5

Q44)

The ratio of momentum of the photons of the 1st and 2nd line of Balmer series of Hydrogen atoms is α/β . The possible values of α and β are:-

Options :

- a. 27 and 20
- b. 3 and 16

c. 5 and 36

d. 20 and 27

Q45)

A LCR series circuit driven with $E_{rms} = 90 \text{ V}$ at frequency $f_d = 30 \text{ Hz}$ has resistance $R = 80 \Omega$, an inductance with inductive reactance $X_L = 20.0 \Omega$ and capacitance with capacitive reactance $X_C = 80.0 \Omega$. The power factor of the circuit is _____.

Options :

a. 0.8

b. 0.64

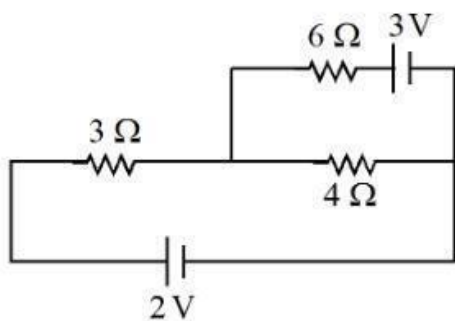
c. 0.9

d. 0.5

Physics Section B

Q46)

Refer to the circuit diagram given below. The heat generated across the 6Ω resistance in 100 second is $\frac{\alpha}{100} \text{ J}$. The value of α is _____. (Nearest integer)



Q47)

An unpolarized light of intensity I_0 passes through polarizer and then through a certain optically active solution and finally it goes to analyser. If the angle between analyser and polariser is 0° and intensity of light emerged from analyser is $\frac{3}{8} I_0$, the angle of rotation of the light by the solution with respect to analyser is _____ degrees.

Q48)

The energy released when $\frac{7}{17.13}$ kg of ${}^7_3\text{Li}$ is converted into ${}^4_2\text{He}$ by proton bombardment is $\alpha \times 10^{32}$ eV. The value of α is _____. (Nearest integer)
(Mass of ${}^7_3\text{Li} = 7.0183$ u, mass of ${}^4_2\text{He} = 4.004$ u, mass of proton = 1.008 u and $1 \text{ u} = 931 \text{ MeV}/c^2$ and Avogadro number = 6.0×10^{23})

Q49)

A three coulomb charge moves from the point $(0, -2, -5)$ to the point $(5, 1, 2)$ in an electric field expressed as $\vec{E} = 2x\hat{i} + 3y^2\hat{j} + 4z\hat{k}$ N/C. The work done in moving the charge is _____ J.

Q50)

A certain gas is isothermally compressed to $\left(\frac{1}{3}\right)^{\text{rd}}$ of its initial volume ($V_0 = 3$ litre) by applying required pressure. If the bulk modulus of the gas is $3 \times 10^5 \text{ N/m}^2$, the magnitude of work done on the gas is _____ J.

RAPTORA

Chemistry Section A

Q51)

An oxide of iron contains 69.9% iron, its empirical formula, is:
(Given : Molar mass of Fe and O are 56 and 16 g mol^{-1} respectively.)

Options :

- FeO
- Fe_2O_3
- Fe_3O_4
- FeO_3

Q52)

If shortest wavelength of hydrogen atom in Lyman series is x , then longest wavelength in Balmer series of He^+ is:

Options :

- a. $\frac{9x}{5}$
- b. $\frac{36x}{5}$
- c. $\frac{x}{4}$
- d. $\frac{5x}{9}$

Q53)

Match the LIST-I with LIST-II

List-I Orbital		List-II Radial nodes and nodal plane	
A.	2s	I.	1 Radial node + two nodal planes
B.	3s	II.	1 Radial node + one nodal plane
C.	3p	III.	2 Radial nodes + No nodal plane
D.	4d	IV.	1 Radial node + No nodal plane

Choose the *correct* answer from the options given below:

Options :

- a. A-IV, B-I, C-III, D-II
- b. A-IV, B-II, C-III, D-I
- c. A-III, B-I, C-IV, D-II
- d. A-IV, B-III, C-II, D-I

Q54)

The pairs among

$A = [\text{SO}_3^{2-}, \text{CO}_3^{2-}]$, $B = [\text{O}_2^{2-}, \text{F}_2]$, $C = [\text{CN}^-, \text{CO}]$, $D = [\text{NH}_3, \text{H}_3\text{O}^+]$ and $E = [\text{MnO}_4^{2-}, \text{CrO}_4^{2-}]$ that do not have similar Lewis dot structure are

Options :

- a. A, B and E
- b. A and E
- c. B, C and D
- d. C and D

Q55)

Arrange the following isothermal processes in order of the magnitude of the work ($p - V$) involved between states 1 and 2.

- A. Expansion in single stage w_A
- B. Expansion in multi stages w_B
- C. Compression in single stage w_C
- D. Compression in multi stages w_D

Choose the correct option.

Options :

- a. $|w_B| > |w_A| > |w_C| > |w_D|$
- b. $|w_C| > |w_D| > |w_A| > |w_B|$
- c. $|w_C| > |w_D| > |w_B| > |w_A|$
- d. $|w_B| > |w_A| > |w_D| > |w_C|$

Q56)

When 0.25 moles of a non-volatile, non-ionizable solute was dissolved in 1 mole of a solvent the vapor pressure of solution was $x\%$ of vapor pressure of pure solvent. What is $x\%$?

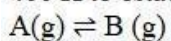
Options :

- a. 50%

- b. 60%
- c. 70%
- d. 80%

Q57)

One mole each of He and A(g) are taken in a 10 L closed flask and heated to 400 K to establish the following equilibrium.



K_c for this reaction at 400 K is 4.0. The partial pressures (in atm) of He and B(g) are respectively (at equilibrium)

(Assume He, A(g) and B(g) behave as ideal gases)

(Given : $R = 0.082 \text{ L atm K}^{-1} \text{ mol}^{-1}$)

Options :

- a. 3.28, 2.624
- b. 2.624, 3.28
- c. 3.28, 0.656
- d. 0.656, 6.56

Q58)

Consider the following data.

Electrolyte	$\Lambda_m^\circ (\text{S cm}^2 \text{ mol}^{-1})$
BaCl ₂	x_1
H ₂ SO ₄	x_2
HCl	x_3

BaSO₄ is sparingly soluble in water. If the conductivity of the saturated BaSO₄ solution is $x \text{ S cm}^{-1}$ then the solubility product of BaSO₄ can be given as

(Here $\Lambda_m = \Lambda_m^\circ$)

Options :

$$\frac{10^6 x^2}{a^2 (x_1 + x_2 - 2x_3)^2}$$

- a.

$$\frac{x^2}{(x_1 + x_2 - 2x_3)^2}$$

b.

$$\frac{a^2(x_1 + x_2 - 2x_3)^2}{10^6 x^2}$$

c.

$$\frac{x^2}{(x_1 + x_2 + 2x_3)^2}$$

d.

Q59)

Given below are two statements:

Statement I: Aluminium is more electropositive than thallium as the standard electrode potential value of $E^\circ_{\text{Al}^{3+}/\text{Al}}$ is negative and $E^\circ_{\text{Tl}^{3+}/\text{Tl}}$ is positive.

Statement II: The sum of first three ionization enthalpies of boron is very high when compared to that of aluminium. Due to this reason boron forms covalent compounds only and aluminium forms Al^{3+} ion.

In the light of the above statements, choose the **correct** answer from the options given below

Options :

- a. Both Statement I and Statement II are true
- b. Both Statement I and Statement II are false
- c. Statement I is true but Statement II is false
- d. Statement I is false but Statement II is true

Q60)

The **correct** statements among the following are.

- A. Basic vanadium oxide is used in the manufacture of H_2SO_4 .
- B. The spin-only magnetic moment value of the transition metal halide employed in Ziegler-Natta polymerization is 2.84 BM.
- C. The p-block metal compound employed in Ziegler-Natta polymerization has the metal in +3 oxidation state.
- D. The number of electrons present in the outer most 'd' orbital of metal halide employed in Wacker process is 8.

Choose the correct answer from the options given below:

Options :

a.

A and B Only

b.

A, C and D Only

c.

C and D Only

d.

B, C and D Only

Q61)

Match the **LIST-I** with **LIST-II**

List-I Electronic configuration of tetrahedral metal ion	List-II Crystal Field Stabilization Energy (Δ_t)
A. d^2	I. - 0.6
B. d^4	II. - 0.8
C. d^6	III. - 1.2
D. d^8	IV. - 0.4

Choose the **correct** answer from the options given below:

a.

A-III, B-IV, C-II, D-I

b.

A-III, B-I, C-IV, D-II

c.

A-III, B-IV, C-I, D-II

d.

A-II, B-I, C-IV, D-III

Q62)

Which of the following are true about the energy of the given d-orbitals tetrahedral complex?

- A. $d_{xy} = d_{xz} > d_{x^2-y^2}$
- B. $d_{xy} = d_{yz} > d_z^2$
- C. $d_{x^2-y^2} > d_z^2 > d_{xz}$
- D. $d_{x^2-y^2} = d_z^2 < d_{xz}$

Choose the correct answer from the given below:

Options :

a.

A, B and D only

b.

A and B only

c.

B and D only

d.

B, C and D only

Q63)

R_F value for 2-methylpropene in a solvent system (Ethyl acetate + ether) is 0.42.

2-methylpropene is treated with dilute H_2SO_4 to give major organic product (X).

R_F value for (X) in the same solvent system under identical condition will be:

Options :

a.

0.42

b.

0.82

c.

0.62

d.

0.12

Q64)

Given below are two statements:

Statement I: 2,6-diethylcyclohexanone and 6-methyl-2-n-propylcyclohexanone are metamers.

Statement II: 2,2,6,6-tetramethylcyclohexanone exhibits keto-enol tautomerism.

In the light of the above statements, choose the *correct* answer from the options given below

Options :

- a. Both Statement I and Statement II are true
- b. Both Statement I and Statement II are false
- c. Statement I is true but Statement II is false
- d. Statement I is false but Statement II is true

Q65)

Given below are two statements:

Statement I: Methane can be prepared by decarboxylation of sodium ethanoate, Kolbe's electrolysis of sodium acetate and reaction of CH_3MgBr with water.

Statement II: Methane cannot be prepared from unsaturated hydrocarbons and by Wurtz reaction.

In the light of the above statements, choose the *correct* answer from the options given below

Options :

- a. Both Statement I and Statement II are true
- b. Both Statement I and Statement II are false
- c. Statement I is true but Statement II is false
- d. Statement I is false but Statement II is true

Q66)

Given below are two statements:

Statement I: 3-phenylpropene reacts with HBr and gives secondary alkyl bromide having a chiral carbon atom as the major product.

Statement II: Aryl chlorides and aryl cyanides can be prepared by Sandmeyer reaction as well as Gattermann reaction.

In the light of the above statements, choose the **correct** answer from the options given below

Given below are two statements:

Statement I: 3-phenylpropene reacts with HBr and gives secondary alkyl bromide having a chiral carbon atom as the major product.

Statement II: Aryl chlorides and aryl cyanides can be prepared by Sandmeyer reaction as well as Gattermann reaction.

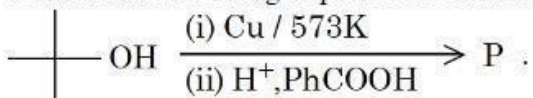
In the light of the above statements, choose the **correct** answer from the options given below

Options :

- a. Both Statement I and Statement II are true
- b. Both Statement I and Statement II are false
- c. Statement I is true but Statement II is false
- d. Statement I is false but Statement II is true

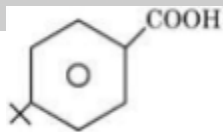
Q67)

Consider the following sequence of reactions

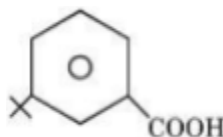


The major product P is:

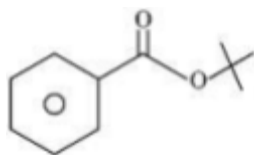
Options :



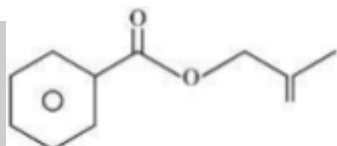
a.



b.



c.



d.

Q68)

Arrange the following compounds according to increasing order of boiling points.

n - C₄H₉OH (A), n - C₄H₉NH₂ (B), n - C₄H₁₀ (C) and C₂H₅NHC₂H₅ (D).

Options :

a.

C < B < A < D

b.

D < C < B < A

c.

C < D < B < A

d.

D < B < A < C

Q69)

Match the **LIST-I** with **LIST-II**

List-I Deficiency Disease		List-II Vitamin	
A.	Scurvy	I.	Pyridoxine
B.	Convulsions	II.	Vitamin A
C.	Cheilosis	III.	Ascorbic Acid
D.	Xerophthalmia	IV.	Riboflavin

Choose the **correct** answer from the options given below:

Options :

a.

A-I, B-III, C-II, D-IV

b.

A-I, B-III, C-IV, D-II

c.

A-III, B-I, C-IV, D-II

d.

A-III, B-I, C-II, D-IV

Q70)

Match the **LIST-I** with **LIST-II**

List-I Amino acid		List-II Positive reaction/Test for functional group present in side chain of amino acid	
A.	Glutamine	I.	Hinsberg's test
B.	Lysine	II.	Neutral FeCl_3 test
C.	Tyrosine	III.	Ceric ammonium nitrate test
D.	Serine	IV.	Hoffman bromamide degradation

Choose the **correct** answer from the options given below:

Options :

a.

A-IV, B-II, C-I, D-III

b.

A-IV, B-I, C-II, D-III

c.

A-III, B-II, C-I, D-IV

d.

A-IV, B-I, C-III, D-II

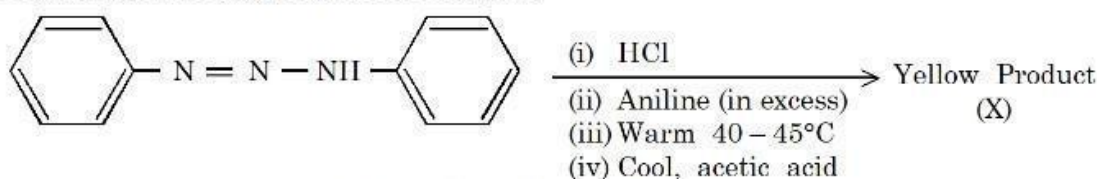
Chemistry Section B

Q71)

First and second ionization enthalpies of lithium are 520 kJ mol^{-1} and 7297 kJ mol^{-1} respectively. Energy required to convert 3.5 mg lithium (g) into $\text{Li}^{2+}(\text{g})$ [$\text{Li}(\text{g}) \rightarrow \text{Li}^{2+}(\text{g})$] is _____ kJ mol^{-1} . (nearest integer)
[Molar mass of Li = 7 g mol^{-1}]

Q72)

Consider the following sequence of reactions.



The percentage of nitrogen in the yellow product (X) formed is _____ %.
(Nearest Integer)

(Given Molar mass in g mol^{-1} H:1, C:12, N:14)

Q73)

4.7 g of phenol is heated with Zn to give product X. If this reaction goes to 60% completion then the number of moles of compound X formed will be _____ $\times 10^{-2}$. (Nearest Integer)

(Given molar mass in g mol^{-1} : H:1, C:12, O:16)

Q74)

Sucrose hydrolyses in acidic medium into glucose and fructose by first order rate law with $t_{1/2} = 3$ hour. The percentage of sucrose remaining after 6 hours is _____. (Nearest integer)

(Given : $\log 2 = 0.3010$ and $\log 3 = 0.4771$)

Q75)

Consider the reaction $\text{X} \rightleftharpoons \text{Y}$ at 300 K. If ΔH^θ and K are $28.40 \text{ kJ mol}^{-1}$ and 1.8×10^{-7} at the same temperature, then the magnitude of ΔS^θ for the reaction in $\text{J K}^{-1} \text{ mol}^{-1}$ is _____. (Nearest integer)

(Given : $R = 8.3 \text{ J K}^{-1} \text{ mol}^{-1}$, $\ln 10 = 2.3$, $\log 3 = 0.48$, $\log 2 = 0.30$)

ANSWERS

1. B
2. D
3. C
4. A
5. B
6. A
7. D
8. D
9. B
10. B
11. A
12. A
13. B
14. B
15. C
16. B
17. B
18. B
19. B
20. A
21. 4
22. 80
23. 3
24. 1279
25. 3
26. B
27. D
28. D
29. B
30. A
31. D
32. B
33. A
34. C
35. A
36. C
37. C
38. A
39. B
40. B
41. A
42. A



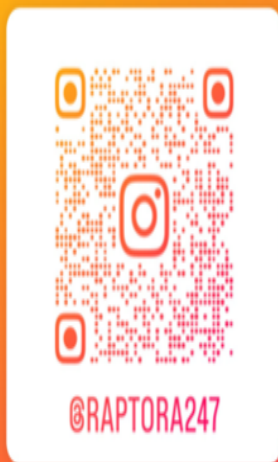
- 43. A
- 44. D
- 45. A
- 46. 3477
- 47. 30
- 48. 6
- 49. 186
- 50. 989
- 51. B
- 52. A
- 53. D
- 54. B
- 55. C
- 56. D
- 57. A
- 58. A
- 59. A
- 60. C
- 61. C
- 62. A
- 63. D
- 64. C
- 65. D
- 66. C
- 67. C
- 68. C
- 69. C
- 70. B
- 71. 3909
- 72. 21
- 73. 3
- 74. 25
- 75. 34



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